## Determination of organic content

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## TYPES OF ORGANIC MATTER

- The total organic matter in soil can be divided into two main groups.
- Soil organic matter= Undecomposed material(I) + Decomposition products (II +III)
- Group I- It consist dead materials, undestroyed debris, waste products of vegetable and animal origin , including living and dead micro-organism. The chemical substances I include cellulose, pectin and other carbohydrate liguin, protein, fats, waxes , resine, etc which are mostely colorless or faintly coloured. They are decomposed by the action of water air and micro-organism.

Group II-known as non-humus constituent of humus extract.simple amino acids,alcohols, aldehydes,and organic bases

 Group-III-dark colored , amorphous substance of non –humous , humic acid Direct method for determination of organic matterextraction with Alkali

 Soil is treated with ammonia or alkali hydroxide and the content of humus is dtermined by either colorimetry or gravimetric method.

- reagents-
- 1N potassium dichromate- dissolve 49.04 gm of reagent grade of K2Cr2O7 in water and dilute to 1 liter.
- 0.5 N ferrous sulphate- dissolve 140 gh of reagrnt grade FeSO4.7H2O in water and add 40 ml of concentrate H2SO4, cool and dilute to 1 liter. Standerised the reagents by titrating it with 10 ml of N potassiudichromate
- Barium Diphenylaminesulphonate- prepare a 0.16 percent acquos solution.

- O- phenantroline ferrous complex-prepare 0.025 M sol. Of phenantroline ferrous complex indicator.
- Sulphuric acid; not less than 96 percent.
- Phosphoric acid: 85 percent
- Procedure-
- Take weighed amount of soil with organic matter in a flask, and add 10 ml of potassium dichromate.
- Add 20 ml of conc. Sulphuric acid, swirl vigorously by hand for 1 minute and let the flask on a sheet of asbestos for about 30 min.then add 200 ml. of water, 10 ml of phosphoric acid and 0.5 ml of barium diphenylaminosulphate indicator.

## • Titration

- Add the ferrous sulphate solution untill the solution is purle, then add the ferrous sulphate in a small amount untill the color turns to green.
- Add 0.5 ml of N potassium dichromate to restore an excess of dichromate and complete the titration by adding ferrous sulphate to drop by drop to a little green point. If more than 8 ml of the available potassium dichromate is reduced, the determination should be repeated with the less soil.

• Percentage of organic matter in soil sample= (mililiter of 1N K2Cr2O7 reduced) x 0.69

weight of sample(gm)

- <u>**REFERRENCES</u>**-Method of soil analysis for soil fertility investigation</u>
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